Notebook, N-02-02-16

This undated notebook was used by Edison, possibly in 1902. It contains lists of chemical compounds intended for an unspecified project, possibly thermocouple or etheric force experiments. The pages are unnumbered. Approximately 30 pages have been used.

Examined by Otto Rohr
10 3 26

Sympatato of Soda 4 grammes
Sulphide lime, 1/4 1/2 1 2 4 grms
Sulphide Potash " "
Sulphide Strontia, " "
Sulphide Barium " "
Sulphide Zinc,
Sulphide Magnesium
Sulphide Aluminum
Sulphide Lithia

Tungstate Lime and 4 grms

Sulphide Potash $\frac{1}{4}$ $\frac{1}{2}$ 1 2 4 7 grms

Sulphide Lime, "

Sulphide Strontia "

Sulphide Barium 1

Sulphide Zinc 1

Sulphide Magnesium 1

Sulphide Aluminum 1

Sulphide Lithia

Tungstate lime 4 grms

Thorium oxide, ¼ ½ 1 2 +4 grms

Tungstic acid 4 grms

Thorium oxide ¼ ½ 1 2 +4 grms

Silicate Lime 4 alone

Silicate Lime 4 grms

Tungstate Soda ¼ ½ 1 2 4 4 grms
Tungstate Lime " "
Tungstate Strontia " "
Tungstate Barium " "

Silicate Lime, 4 grms

Sulphide Lime ¼ ½ 1 2 4 3 grms
Sulphide Barium "
Sulphide Strontia "
Sulphide Zinc "
Sulphide Lithia "
Oxide Thorium "

Silicate Zinc 4 grms +

Silicate Lime ¼ ½ 1 2 + 4 grm
" Tin "
" Bismuth "
" Lead "
" Iron "
" Cobalt "
" Nickel "
" Antimony "
" Strontia "
" Baryta "
" Alumina "
" Magnesia "
" Copper "
" Manganese "
Zinc
Cadmium

Silicate Barium 4 grms

Silicate Lime ¼ ½ 1 2 4 4 grms

Tin

Bi

Pb

Fe

Co

Ni

Sb

Sr

Ba

Al

Mg

Cu

Mn

Ur

W

Mo

Cr

Zn

Cd

Silicate Strontia 4 grms
and
Silicate Tin ¼ ½ 1 2 4 grms
Bi
Pb
Fe
Co
Ni
Sb
Sr
Ba
Al
Mg
Cu
Mn
U
Mo
W
Cr
Li
K
Na
Zn
Cd

Fluoride Lime 4 grms

Sulphide Sn ¼ ½ 1 2 4 grms
Pb
Fe
Co
Ni
Sb
Sr
Ba
Al
Mg
Cu
Mn
Ur
Mo
W
Cl
Li
K
Na
Bi
Zn
Cd

Fluoride Lime 4 grms a—

Fluoride.	Sn	¼	½	1	2 a 4 grs
	Pb				
	Fe				
	Co				
	Ni				
	Sb				
	Sr				
	Ba				
	Al				
	Mg				
	Cu				
	Mn				
	Ur				
	Mo				
	W				
	Ag				
	K				
	Na				
	Bi				
	Zn				
	Cd				

Sulphate of Jm 4 grms ↵

Sulphate of Pb ½ 1 2 & 4 grms

Fe
Co
Ni
Sb
Sr
Ba
Al
Mg
Cu
Mn
Ur
Mo
Li
Na
Ca
Cr
Bi
Zn
Cd

Sulphate of lead 4 grms

Sulphate of Fe 4 ½ 1 2 4 grs

Co

Ni

Sb

Sr

Ba

Al

Mg

Cu

Mn

Ur

Mo

Li

Na

Ca

Cr

Zn

Cd

Sulphate Fe 4 grms +

Sulphate Co
Sb
Ni
Sr
Ca
Ba
Al
Mg
Cu
Mn
Ur
Mo
Li
Na
Cr
Zn
Cd

Sulphate of Nickel 4 grms

Sulphate of Co
Sr
Ca
Ba
al
Mg
Cu
Mn
Uv
Mo
Cr
Li
Na
Bi
Sb
Fe
Zn
Cd

Sulphate Cobalt 4 grms &

Sulphate. Sr ¼ ½ 1 2 & 4 grm

Ca

Ba

Al

Mg

Cu

Mn

Ur

Mo

Cr

Li

Na

Bi

Sb

Zn

Cd

Sulphate Lime 4 groups

Sulphate.	Sr	4	1/2	1	2	4
	Ba					
	Al					
	Mg					
	Cu					
	Mn					
	Ur					
	Mo					
	Cs					
	Li					
	Na					
	Bi					
	Sb					
	Zn					
	Cd					

Sulphate Barium 4 grms &

Sulphate of Mg ¼ ½ 1 2 & 4 grms

Al

Mg

Cu

Mn

U

Mo

Cr

Li

Na

Bi

Sb

Zn

Cd

Sulphate Aluminum 4 grms 4

Sulphate Mg 7 2 1 2 4 4

Cu

Mn

Ur

Mo

Cr

Li

Na

Bi

Sb

Zn

Cd

Sulphate Magnesium 4 grms. ℞

Sulphate Cu ¼ ½ 1 2 ℞ 4

Mn
Ur
Mo
Cr
Al
Na
Bi
Sb
Zn
Cd

Sulphate Manganese 4 grms ♀

Sulphate	Ur	¼	½	1	2	4
	Mo					
1	Cr			1		
	Li					
1	Na			1		
	Bi					
1	Sb			1		
	Zn					
1	Ca					

Sulphate Uranium 4 grms ⊕

Sulphate	Mo	¼	½	1	2	4
	Cr				1	
	Li					
	Na					
	Bi				1	
	Sb					
	Zn				1	
	Cd					

Sulphate Chromium 4 grms &

Sulphate Li ¼ ½ 1 2 & 4
Na 1
Bi
Sb 1
Zn
Ca 1

Sulphate Lithia 4 grms

Sulphate Na $\frac{1}{4}$ $\frac{1}{2}$ 1 2 + 4

Bi

Sb 1

Zn 1

Cu 1

Sulphate Soda 4 grms &

Sulfate Bi ¼ ½ 1 2 & 4
Sb /
Zn
Cd

Sulphate Bismuth 4 gms

Sulphate Sb ¼ ½ 1 2 4
Zn
Cd

Sulphate Zinc 4 grms &
Sulphate Cd ¼ ½ 1 2 & 4

Phosphate Lithea 4 grms

Phos Uranium 4 grms

Phos Zinc " "

Phos Barium "

Phos Strontium "

Arsenite Barium "

Borate of Zinc "